M. L. Wolfrom, J. E. Mahan, P. W. Morgan and G. F. Johnson. Osage Orange Pigments. VI. Isoflavone Nature of Osajin.

Page 1251. Column 2, line 3 from the end, for "$C_{26}H_{34}O_5(CH_3CO)$" read "$C_{26}H_{33}O_5(CH_3CO)$."—M. L. WOLFROM.

M. L. Wolfrom and J. E. Mahan. Osage Orange Pigments. VII. Isoflavone Nature of Pomiferin.

Page 1253. In footnote (1), for "2481" read "1248."—M. L. WOLFROM.

W. T. Haskins, Raymond M. Hann and C. S. Hudson. Synthesis of the Epimer of Cellobiose (4-[β-D-Glucopyranosido]-D-mannose).

Page 1725. In Formula I, bottom of left side, the C atom should have H_2 attached, as in Formulas II and III.—W. T. HASKINS, R. M. HANN and C. S. HUDSON.

Arthur C. Cope, Kathryn E. Hoyle and Dorothea Heyl. The Rearrangement of Allyl Groups in Three-Carbon Systems. I.

Pages 1846–1847. By oversight in the Editor's office, the titles of the figures were omitted.

Fig. 1.—Rate of rearrangement of (1-alkenyl)-allylmalononitriles at 135.7 ± 0.5°: ○, rearrangement of XIII; ◒, XV.

Fig. 2.—Rate of rearrangement of ethyl (*s*-1-alkenyl)-allylcyanoacetates at 176.5 ± 0.5°: ◒, rearrangement of VII; ○-, V; ◑, I; ◐, III; ◔, XI; ○, IX.

Fig. 3.—Rate of rearrangement of ethyl (primary 1-alkenyl)-allylmalonates at 193.1 ± 0.5°: ◒, rearrangement of XVII; ○, XIX.

Henri A. Levy and Robert B. Corey. The Crystal Structure of *dl*-Alanine.

Page 2097. In the tenth line from the bottom, for "$(1 + \cos^2 2\theta)/\sin 2\theta \sin \phi$" read "$(1 + \cos^2 2\theta)/\cos^2 \mu \sin \phi$" in which μ is the angle of latitude of the reflection. This correct expression was used in the calculation of the *F* values published in this article upon which the determination of the structure of *dl*-alanine was based.—HENRI A. LEVY and ROBERT B. COREY.

Raymond M. Hann and C. S. Hudson. D-Galactosan <1,5>β<1,3>, a New Anhydride of D-Galactose.

Page 2242. In Column 2, lines 5 and 6, for "2-(D-galactohexahydroxyhexyl)-benzimidazole" read "2-(D-galactopentahydroxypentyl)-benzimidazole."—R. M. HANN and C. S. HUDSON.

David B. Sprinson. The Synthesis of N-Substituted Choline Carbamates and Trimethyl-β-phenylaminoethylammonium Chloride.

Page 2251. In Table III, the melting point of the sixth compound should be "97–99°" instead of "87–99°."

George F. Davies and E. C. Gilbert. Heats of Combustion and of Formation of the Nine Isomeric Heptanes in the Liquid State.

Page 2731, Table I, the boiling point for 2,4-dimethylpentane should be "80.49" instead of "90.49."—E. C. GILBERT.

F. F. Blicke and M. F. Zienty. Acid Amides as Hypnotics. III. Disubstituted Acetamides.

Page 2780. In Table I, the heading "Malonic ester" is spaced in such a way that it might be assumed that the melting points refer to the malonic esters, while actually they are for the malonic acids.—F. F. BLICKE.

Lloyd M. Cooke, Joseph L. McCarthy and Harold Hibbert. Studies on Lignin and Related Compounds. LXI. Hydrogenation of Ethanolysis Fractions from Maple Wood (Part 2).

Page 3059. In column 1, line 25, for "—C—O—C—" read "—C—C—C—".—HAROLD HIBBERT.

Homer E. Stavely. Molecular Rearrangements of 17-Hydroxy-pregnane Compounds.

Page 3128. In Formula III, the first (upper) small number 17 should be 17a.—HOMER E. STAVELY.

J. L. Bullock and E. T. Mitchell. The Vapor Phase Nitration of Toluene.

Page 3231. In the last data line of Table I, for "150" read "250."—E. T. MITCHELL.

Philip G. Stevens and James H. Richmond. The Mechanism of Elimination Reactions. I. The Decomposition of Quaternary Ammonium Bases and of Xanthate Esters.

Page 3132. The formula equation given in column 1 should be a part of footnote (6).—PHILIP G. STEVENS.